2000 Vol. 2, No. 13 1799–1801

## A Facile Route to Indolo[2,1-a]isoquinolines and Dibenzopyrrocoline Alkaloids

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Received March 14, 2000

## **ABSTRACT**

$$R^{1}$$
 $R^{2}$ 
 $R^{3}$ 
 $R^{4}$ 
 $R^{2}$ 
 $R^{4}$ 
 $R^{2}$ 
 $R^{4}$ 
 $R^{2}$ 
 $R^{2}$ 
 $R^{4}$ 
 $R^{2}$ 
 $R^{4}$ 
 $R^{5}$ 
 $R^{5}$ 
 $R^{5}$ 

Treatment of 1-(2'-bromobenzyl)-3,4-dihydroisoquinolines 2 in the presence of  $K_2CO_3$  in boiling DMF efficiently provided a variety of alkoxy-substituted indolo[2,1-a]isoquinolines 3. Application of this cyclization to 7-benzyloxyisoquinoline derivatives, followed by further elaboration of the resultant 2-benzyloxy-5,6-dihydroindolo[2,1-a]isoquinolines 16a,b, led to the formal synthesis of dibenzopyrrocoline alkaloids,  $(\pm)$ -cryptaustoline (1a) and  $(\pm)$ -cryptowoline (1b).

Indolo[2,1-a]isoquinoline has a unique nitrogen-containing tetracyclic structure, characteristic of dibenzopyrrocoline alkaloids, cryptaustoline **1a** and cryptowoline **1b**, isolated

indolo[2,1-α]isoquinoline

1a: cryptaustoline R<sup>1</sup>=R<sup>2</sup>=OMe

1b: cryptowoline R<sup>1</sup>+R<sup>2</sup>=OCH<sub>2</sub>O

from the bark of *Cryptocarya bowiei*.<sup>1,2</sup> Several methods for construction of this structure,<sup>3–8</sup> including the well-known benzyne reaction<sup>3</sup> or oxidative coupling<sup>4</sup> of 1-benzylisoquinolines, have been reported. Antileukemic and antitumor activities of such bases have been reported, and their ammonium salts have been expected to enhance the activities.<sup>31,9</sup>

As shown in Scheme 1, we initially encountered this structure in the intramolecular cyclization products of *erythro*-1-[(2'-bromopheny)hydroxymethyl]-1,2,3,4-tetrahydroisoquinolines **5**, which were prepared in three more

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Scheme 1. Preceding Studies<sup>10–12</sup>

steps of reactions from 2'-bromobenzyl-3,4-dihydroisoquinolines 2, in the presence of  $K_2CO_3$ . <sup>10,11</sup> Recently, we reported  $Bu_3SnH$ -induced aryl radical cyclization, which competitively gave 5,6-dihydroindolo[2,1-a]isoquinolines 3 and aporphines 4 from 2. <sup>12</sup> In the present study, we developed a convenient method for selective preparation of a variety of alkoxy-substituted indolo[2,1-a]isoquinolines 3 from the same substrates 2, involving 3a-d, which cannot be obtained by the benzyne method noted above. <sup>3</sup>

**Scheme 2.** Synthesis of 5,6-Dihydroindolo[2,1-*a*]isoquinolines

entry isolated yields and mps of 3

R<sup>5</sup>=H

**2a**:  $R^1$ = $R^2$ = $R^3$ = $R^4$ =OMe 95 % mp 193-195 °C (A) **2b**:  $R^1$ = $R^2$ =OMe,  $R^3$ + $R^4$ =OCH<sub>2</sub>O 92 % mp 198-200 °C (B) **2c**:  $R^1$ + $R^2$ =OCH<sub>2</sub>O,  $R^3$ = $R^4$ =OMe 78 % mp 177-180 °C (B) **2d**:  $R^1$ + $R^2$ = $R^3$ + $R^4$ =OCH<sub>2</sub>O 95 % mp 205-206 °C (A)

 $R^3=H$ **2e**: R<sup>1</sup>=R<sup>2</sup>=R<sup>3</sup>=R<sup>4</sup>=OMe mp 207-208 °C (C]<sup>a</sup> 95 % **2f**:  $R^1 = R^2 = OMe$ ,  $R^3 + R^4 = OCH_2O$  95 % mp 241-242.5 °C (C) **2g**:  $R^1 + R^2 = OCH_2O$ ,  $R^3 = R^4 = OMe$ 89 % mp 212-216 °C (C) **2h**:  $R^1 + R^2 = R^3 + R^4 = OCH_2O$ 79 % mp 214-217.5 °C (C) 2 i: R<sup>1</sup>=R<sup>2</sup>=OMe, R<sup>4</sup>=R<sup>5</sup>=H 83 % mp 177.5-179.5 °C (B) **2 j**:  $R^1 + R^2 = OCH_2O$ ,  $R^4 = R^5 = H$ 76 % mp 189-193.5 °C (B)

Solvents for crystallization: A, EtOH; B, MeOH-CH $_2$ Cl $_2$ ; C, MeOH-Et $_2$ O

a: lit. 1 mp 199 °C; 4a 201-203 °C; 3d 202-203 °C; 3a 202-204 °C; 8,10 204-205 °C; 3g,i 209-210 °C When substrates **2a**–**d** with an alkoxy group at their 3′ position or substrates **2e**–**j** without the alkoxy group, including the compounds **2i** and **2j** having no substituent on the phenyl group except for a Br atom, were heated in the presence of 2 mol equiv of K<sub>2</sub>CO<sub>3</sub> in boiling DMF for 3 days (Scheme 2), the corresponding 5,6-dihydroindolo[2,1-*a*]isoquinolines **3** were obtained in 76–95% isolated yields by crystallization. Dimeric products at their C-12 position were not detected at all. <sup>3g,4c,7b</sup> The cyclization did not proceed without alkali. K<sub>3</sub>PO<sub>4</sub> worked well, similarly to K<sub>2</sub>CO<sub>3</sub>, but stronger bases such as BuLi and KO'Bu, did not. Replacement of DMF with DMSO resulted in the formation of a black tar.

Under the same conditions, the 2'-iodo derivative **6a** was consumed much faster than the bromide **2a**, and the cyclization finished within 1.5 days. However, its 2'-methoxy derivative **6b** was recovered unchanged. The cyclization of a readily accessible 3'-bromopapaverine **7**<sup>13</sup> also proceeded smoothly to give a fully aromatized indolo[2,1-*a*]isoquinoline **8**, mp 225.5–228.0 °C (MeOH, lit.<sup>3i</sup> mp 210 °C), within 2 days quantitatively (Scheme 3), although DDQ oxidation of **3e** gave **8** (65%).

**Scheme 3.** Synthesis of Indolo[2,1-*a*]isoquinoline **8** 

MeO 
$$\times$$
 X  $\times$  2CO3 DMF reflux N2 MeO  $\times$  MeO

It had been reported that the thermal electrocyclization of a pentadienyl anion gave a cyclopentenyl anion.<sup>14</sup> We had assumed that such an effect might have contributed to these cyclizations (Scheme 4), until the following fact was disclosed. 1-Benzyldihydroisoquinoline 9, having two methyl

**Scheme 4.** An Assumed Thermal Electrocyclization Process<sup>12</sup>

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groups at the benzyl position, also underwent a similar cyclization with loss of one methyl group to give 12-methylindolo[2,1-a]isoquinoline **10**, mp 224–226 °C (MeOH–CH<sub>2</sub>Cl<sub>2</sub>, lit.  $^9$  216–217 °C) (Scheme 5). The dehydro deriva-

**Scheme 5.** Formation of Indolo[2,1-*a*]isoquinoline **10** 

tive (11) did not cyclize and was recovered unchanged. The dihydro derivatives of 2e, 1,2,3,4-tetrahydroisoquinolines (12, 13),<sup>11</sup> were also recovered unchanged. An attempt to produce a quinoline ring using the homologue 14 failed, and it was recovered unchanged. On the basis of these results, a reaction pathway via i, ii, and iii (Scheme 6) which starts with a nucleophilic addition of the isoquinoline nitrogen and ends up in the formation of a stable conjugated system, "indole ring", was proposed for this versatile cyclization.

Application of this cyclization on 7-benzyloxy-3,4-dihydroisoquinoline  ${\bf 15a}^{3a}$  and  ${\bf 15b}^{3b}$  gave 5,6-dihydroindolo[2,1-

**Scheme 6.** A Probable Pathway to Indolo[2,1-a]isoquinolines **3. 8.** and **10** 

*a*]isoquinolines **16a**, mp 146–148 °C (MeOH), and **16b**, mp 159–161 °C (MeOH, lit.<sup>3b</sup> mp 157–158 °C), almost quantitatively (82% and 83% isolated yields by crystallization). Dihydroisoquinolines **16a** and **16b** were further converted by treatment with excess NaBH<sub>3</sub>CN in AcOH almost quantitatively to air-sensitive tetrahydroindolo[2,1-*a*]isoquinolines **17a** and **17b**, respectively (Scheme 7). In

**Scheme 7.** Synthesis of Dibenzopyrrocoline Alkaloids Cryptaustoline **1a** and Cryptowoline **1b** 

view of the previous conversion of these compounds into  $(\pm)$ -cryptaustoline (1a) and  $(\pm)$ -cryptowoline (1b),  $^{3a,b,i,7b}$  this constitutes a formal synthesis of the alkaloids.

**Supporting Information Available:** Characterization data for products **3**, **8**, **10**, and **16**. This material is available free of charge via the Internet at http://pubs.acs.org.

OL005802D

Org. Lett., Vol. 2, No. 13, **2000** 

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